ORALLY ACTIVE CEPHALOSPORINS

III. SYNTHESIS AND STRUCTURE-ACTIVITY RELATIONSHIPS OF NEW 3-HETEROCYCLICTHIOMETHYLTHIO-7β-[(Z) 2-(2-AMINOTHIAZOL-4-YL)-2-HYDROXYIMINOACETAMIDO] 3-CEPHEM-4-CARBOXYLIC ACIDS

MASAHARU KUME, TADATOSHI KUBOTA, YASUO KIMURA, HIROMU NAKASHIMIZU and KIYOSHI MOTOKAWA

Shionogi Research Laboratories, Shionogi & Co., Ltd., Sagisu, Fukushima-ku, Osaka 553, Japan

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3-Heterocyclicthiomethylthio- 7β -[(Z)-2-(2-aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3cephem-4-carboxylic acids (2) were synthesized. Their antibacterial activity and oral absorbability were much influenced by the structure of heteroaromatic moiety in the side chain at the 3-position of a cephem nucleus. In this cephalosporin series, 3-thiadiazolylthiomethylthiocephalosporins (2k, 2l and 2m) exhibited potent antibacterial activity against both Gram-positive and Gram-negative bacteria, whereas 3-(2-methyl-1,2,3-triazol-4-yl)thiomethylthiocephalosporin (2b) and 3-(pyridin-2yl)thiomethylthiocephalosporin (2n) showed better oral absorption in mice than the other cephalosporins prepared. The structure-activity relationships of 2 are presented.

In our preceding paper,¹⁾ we have reported on the synthesis, the antibacterial activity and the oral absorbability of 7β -[(Z)-2-(2-aminothiazol-4-yl)-2-hydroxyiminoacetamido]cephalosporins with 1,2,3-triazole in the C-3 side chain. Among these compounds 3-(1,2,3-triazol-4-yl)thiomethylthio-3-cephem-4-carboxylic acid (1) was selected for further evaluation, which showed well-balanced good antibacterial activity against Gram-positive and Gram-negative bacteria and attained high plasma level after oral administration in mice and monkeys. That study has also given us the finding that the spacer moiety between C-3 of the cephem nucleus and C-4' of 1,2,3-triazole was a factor affecting the oral absorbability;



Fig. 1. 3-Heterocyclicthiomethylthiocephalosporins (1 and $2a \sim 2n$).

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good oral absorbability was obtained when the spacer moiety was a three-atom chain, especially a thiomethylthic chain. In this paper, we wish to report the synthesis of 2 and the effects of the alteration of the heteroaromatic ring in the C-3 side chain of 1 on antibacterial activity and oral absorbability in mice.

Chemistry

Scheme 1 outlines two methods for the synthesis of the 3-heterocyclicthiomethylthiocephalosporins (2). The preparation of $2a \sim 2d$, 2h and 2n was carried out by method A and that of $2e \sim 2g$ and $2i \sim 2m$ by method B.

Method A, which had also been used for the synthesis of compound 1, involves the coupling of 3-methanesulfonyloxycephalosporin¹⁾ (3) with sodium thiolate, prepared *in situ* by methanolysis of the corresponding heterocyclicthiomethyl thioacetate, to give 3-substituted compound 4. This method gave compound 4 in one step but in some cases could not give 4, even in a poor yield, probably due to the



BH = diphenylmethyl, Tr = triphenylmethyl, mCPBA = m-chloroperbenzoic acid, HMPA = hexamethylphosphoric triamide.



instability of the intermediate sodium thiolate.²⁾ An alternative procedure (method B), which does not include the thiolate intermediate, was suited for the synthesis of 4 in such cases. Method B is more general than method A and also applicable to the compounds prepared by method A. In method B, 3 was oxidized with *m*-chloroperbenzoic acid (*mCPBA*) and treated successively with sodium hydrosulfide and silver nitrate to yield silver salt (5), which was reacted with heterocyclicthiomethyl iodide in hexamethylphosphoric triamide (HMPA) to afford 3-substituted sulfoxide (6). 3-(4-Methyl-1,2,4-triazol-3-yl)thiomethylthiocephalosporin (6g) was prepared by modified method B. As shown in Scheme 2, 5 was reacted with [1-trityl-1,2,4-triazol-3-yl]thiomethyl iodide (7) to give 3-substituted compound (8). The trityl group was removed with *p*-toluenesulfonic acid to yield compound 9, which was methylated with methyl triflate to give compound 6g. The reduction of sulfoxide (6) with phosphorus trichloride (PCl₃) gave compound 4.

The preparation of thioacetates and iodides used here was shown in Scheme 3. Each isomers ($12 \sim 14$; 18, 19; 28, 29) were assigned by means of NOE or ¹³C NMR study,³⁾ or by comparison with authentic samples.

Finally the deprotection of 4 by the conventional method using aluminum chloride (AlCl₃) produced the desired cephalosporin derivatives $(2a \sim 2n)$.

Antibacterial Activity and Oral Absorption

The *in vitro* antibacterial activity of the new cephalosporins (2) against selected Gram-positive and Gram-negative bacteria is shown in Table 1. Their plasma levels and urinary recovery after oral administration (40 mg/kg) to mice are summarized in Table 2. Also shown is their relative bioavailability which was calculated according to the following equation:

relative bioavailability (%) = $\frac{\text{urinary recovery after po dosage}}{\text{urinary recovery after sc dosage}} \times 100$

and we used this bioavailability and plasma levels as measures of gastrointestinal absorption.

Most of the new cephalosporins exhibited potent activity against the Gram-positive and Gram-negative bacteria except *Pseudomonas aeruginosa* ATCC 25619, irrespective of the heteroaromatic ring. Derivatives with thiadiazole in the C-3 side chain $(2k \sim 2m)$ showed higher activity against a wide range of bacteria than the other cephalosporins prepared, including compound 1. Compounds 2k and 2l showed the most potent activity against Gram-positive bacteria including *Staphylococcus aureus* SR3131 (MRSA) among all of the compounds tested.

Compounds (2b and 2n) having 2-methyl-1,2,3-triazole or pyridine in the C-3 side chain, showed good



Organiam	MIC $(\mu g/ml)$							
Organism	2a	2b	2c	2d	2e	2f	2g	2h
Staphylococcus aureus FDA 209P JC-1	0.2	0.2	0.1	0.2	0.2	0.2	0.2	0.4
S. aureus Smith	0.2	0.2	0.1	0.4	0.2	0.4	0.4	0.8
S. aureus SR3131	6.3	12.5	3.1	3.1	3.1	12.5	12.5	12.5
S. epidermidis ATCC 14990	0.2	0.2	0.1	0.2	0.2	0.4	0.2	0.8
Streptococcus pyogenes C-203	0.006	0.006	0.006	0.006	0.006	0.006	0.006	0.05
S. pneumoniae Type 1	0.02	0.02	0.01	0.02	0.02	0.02	0.02	0.1
Escherichia coli H	0.1	0.2	0.05	0.05	0.1	0.1	0.05	0.05
E. coli NIHJ JC-2	0.4	3.1	0.2	0.2	0.8	0.8	0.1	0.4
E. coli EC-14	0.2	0.8	0.1	0.1	0.2	0.4	0.05	0.1
E. coli SR377	0.8	3.1	0.4	0.8	0.8	3.1	0.4	3.1
Klebsiella pneumoniae SR1	0.2	0.8	0.1	0.05	0.2	0.2	0.05	0.05
Proteus mirabilis PR-4	0.1	0.4	0.05	0.05	0.1	0.1	0.05	0.02
P. vulgalis CN-329	0.4	0.2	0.05	0.1	0.2	0.8	0.2	0.2
Morganella morganii SR9	0.2	0.8	0.1	0.2	0.4	0.8	0.1	0.4
Enterobacter cloacae SR233	0.8	3.1	0.4	0.8	0.8	1.6	0.4	12.5
Serratia marcescens ATCC 13880	3.1	12.5	1.6	3.1	6.3	12.5	3.1	12.5
Pseudomonas aeruginosa ATCC 25619	>100	>100	>100	>100	>100	>100	>100	>100

Table 1. In vitro antibacterial activity of cephalosporins $(2a \sim 2n \text{ and } 1)$.

Organicm	MIC ($\mu g/ml$)							
Organishi	2i	2j	2k	21	2m	2n	1	
Staphylococcus aureus FDA 209P JC-1	0.2	0.2	0.1	0.1	0.1	0.05	0.2	
S. aureus Smith	0.2	0.4	0.1	0.1	0.2	0.2	0.2	
S. aureus SR3131	6.3	6.3	1.6	1.6	3.1	3.1	12.5	
S. epidermidis ATCC 14990	0.2	0.2	0.1	0.1	0.2	0.1	0.2	
Streptococcus pyogenes C-203	0.006	0.006	< 0.003	< 0.003	0.006	0.006	0.006	
S. pneumoniae Type 1	0.02	0.02	0.01	0.01	0.02	0.01	0.02	
Escherichia coli H	0.05	0.2	0.05	0.02	0.1	0.2	0.05	
E. coli NIHJ JC-2	0.4	1.6	0.4	0.2	0.8	1.6	0.2	
E. coli EC-14	0.1	0.8	0.2	0.1	0.4	0.8	0.1	
E. coli SR377	0.4	1.6	0.4	0.4	0.8	1.6	1.6	
Klebsiella pneumoniae SR1	0.1	0.4	0.05	0.05	0.2	0.4	0.1	
Proteus mirabilis PR-4	0.05	0.1	0.05	0.05	0.1	0.2	0.05	
P. vulgalis CN-329	0.2	0.4	0.1	0.1	0.1	0.1	0.2	
Morganella morganii SR9	0.1	0.4	0.05	0.1	0.1	0.8	0.1	
Enterobacter cloacae SR233	0.8	1.6	0.4	0.4	0.8	1.6	0.8	
Serratia marcescens ATCC 13880	1.6	12.5	1.6	1.6	3.1	6.3	3.1	
Pseudomonas aeruginosa ATCC 25619	>100	>100	>100	>100	>100	>100	>100	

oral absorbability in mice as well as 1 but their plasma levels did not reach the level of 1. All of the other compounds exhibited only a limited absorption. The phenomenon that replacement of 1,2,3-triazole in the C-3 side chain with other heteroaromatics decreased oral absorbability, had also been observed in the case of cefatrizine and its congeners.^{4,5)} Furthermore, two of the compounds possessing *N*-methyl-1,2,3-triazole or *N*-methyl-1,2,4-triazole in the C-3 side chain (**2b** or **2e**) were found to exhibit as good oral absorbability as the compounds possessing *N*-unsubstituted 1,2,3-triazole or 1,2,4-triazole (1

Compound	Plasma level (µg/ml)		Urinary	Relative		Plasma level (µg/ml)		Urinary	Relative
	15 minutes	120 minutes	recovery (%)	bioavaila- bility (%)	Compound	15 minutes	120 minutes	recovery (%)	bioavaila- bility (%)
2a	3.73	2.83	5.9	17	2i	0.74	0.30	2.1	6.6
2b	17.2	6.10	8.0	46	2j	2.04	0.70	1.8	8.3
2c	1.82	0.73	1.5	4.0	2k	1.86	0.67	2.7	12
2d	4.78	2.81	5.4	17	21	3.34	0.87	3.4	13
2e	3.36	1.28	9.4	24	2m	1.34	0.31	3.5	13
2f	0.90	0.76	1.5	6.9	2n	18.7	12.8	3.5	73
2g	0.78	0.38	1.2	3.8	1	29.6	51.3	5.5	36
2h	0.80	2.79	0.6	2.5					

Table 2. Plasma levels, urinary recovery and oral bioavailability of cephalosporins (2a ~ 2n, 1) in mice after oral administration of 40 mg/kg.

Mice: ICR-strain, 6-week-old male, n = 5.

The urinary samples were collected over 2 hours and the relative bioavailability was calculated from the urinary recovery over 2 hours.

or 2d), but other N-methylated derivatives (2a, 2c or 2f, 2g) showed lower oral absorbability.

Experimental

MP was determined with a Yanagimoto micro melting point apparatus and uncorrected. IR spectra were taken on a Jasco IR-700 spectrometer. ¹H NMR spectra were recorded at 200 MHz on a Varian VXR-200 NMR spectrometer using TMS or sodium 2,2-dimethyl-2-silapentan-5-sulfonate (in D_2O) as an internal standard. Mass spectra (EI-MS) was measured on a Hitachi M-68 mass spectrometer. The following abbreviations are used: s, singlet; d, doublet; m, multiplet; br, broad; ABq, AB quartet. All reactions under anhydrous conditions were carried out using anhydrous solvents dried over Molecular Sieves type 4A in a nitrogen atmosphere.

Determination of Antibacterial Activity

All the *in vitro* antibacterial activities are given as MIC in μ g/ml required to prevent growth of the bacterial culture. MICs were determined by the serial agar dilution method (Sensitivity Disk Agar-N) after incubation at 37°C for 18~20 hours with an inoculum size of about 10⁶ cells/ml.

Oral Absorption Study

Male ICR-strain mice aged 6 weeks weighing $24 \sim 30$ g were used in groups of 5. The antibiotics were given to mice orally in a single dose of 40 mg/kg or subcutaneously in 20 mg/kg as a solution in dilute aqueous sodium bicarbonate. Plasma samples were collected at 0.25 and 2 hours respectively after dosing and urine specimens were collected over a period of 2 hours after dosing. The concentrations of the test compounds were determined by the Band Culture method⁶ using *Escherichia coli* 7437 as a test organism and Trypto-soy agar as the test medium.

Method A: Diphenylmethyl 7β -[(Z)-2-(2-tert-Butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-(1-methyl-1H-1,2,3-triazol-4-yl)thiomethylthio-3-cephem-4-carboxylate (4a)

A solution of 12 (392 mg, 1.93 mmol) in THF (2 ml) and DMF (6 ml) was cooled to -78° C and treated with a solution of sodium methoxide in MeOH (1.28 N, 1.33 ml). After the mixture was stirred at -78° C for 15 minutes, a solution of 3 (1.50 g, 1.54 mmol) in DMF (5 ml) was added dropwise to the above mixture. After being stirred at the same temperature for 50 minutes, the mixture was neutralized with 10% HCl, diluted with water and extracted with EtOAc. The extract was washed with brine four times, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene - EtOAc, 3:2) to yield 1.43 g (90%) of 4a as colorless froth.

Compounds $4b \sim 4d$, 4h and 4n were similarly prepared from 3 with corresponding thioacetates (13,

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com- pound No.	¹ H NMR (Solvent, δ)	$\frac{\text{IR (CHCl}_3) \text{ cm}^{-1}}{(\text{C}=\text{O})}$
4 a	$(CDCl_3)$; 8.83 (1H, br s), 7.66 (1H, d, $J=8.6$ Hz), 7.50 ~ 7.15 (26H, m), 6.99 (1H, s) 6.93 (1H s) 5.91 (1H dd $J=4.7$ and 8.6 Hz) 5.06 (1H d $J=4.7$ Hz) 4.08	1781, 1714, 1684
4b	(2H, s), 3.92 (3H, s), 3.51 and 3.35 (2H, ABq, $J=16.8$ Hz), 1.49 (9H, s) (CDCl ₃); 8.85 (1H, br s), 7.61 (1H, d, $J=8.3$ Hz), 7.49 (1H, s), $7.45 \sim 7.15$ (25H, m), 6.99 (1H, s), 6.90 (1H, s), 5.86 (1H, dd, $J=4.9$ and 8.3 Hz), 5.07 (1H, d, $J=4.9$ Hz), 4.10 (3H, s), 3.98 (2H, s), 3.43 and 3.32 (2H, ABq, $J=16.8$ Hz), 1.50	1785, 1717, 1686
4c	(9H, s) (CDCl ₃ -CD ₃ OD); 7.73 (1H, s), 7.50~7.15 (25H, m), 7.06 (1H, s), 6.96 (1H, s), 5.99 (1H, d, J =4.8 Hz), 5.12 (1H, d, J =4.8 Hz), 3.95 (3H, s), 3.94 and 3.87 (2H,	1784, 1714, 1683
4d	ABq, J=13.7 Hz), 3.55 and 3.43 (2H, ABq, J=17.3 Hz), 1.52 (9H, s) (CDCl ₃); 8.6~8.4 (1H, br s), 7.89 (1H, s), 7.55~7.05 (41H, m), 7.02 (1H, s), 6.94 (1H, s), 5.90 (1H, dd, J=4.9 and 8.5 Hz), 4.95 (1H, d, J=4.9 Hz), 4.17 (2H, s),	1781, 1715, 1683
4e	3.41 and 3.30 (2H, ABq, $J = 17.4$ Hz), 1.50 (9H, s) (CDCl ₃); 9.3~9.15 (1H, br s), 7.84 (1H, s), 7.83 (1H, d, $J = 8.6$ Hz), 7.5~7.2 (25H, m), 6.95 (1H, s), 6.93 (1H, s), 5.91 (1H, dd, $J = 4.6$ and 8.6 Hz), 5.01 (1H, d, $J = 4.6$ Hz), 4.48 and 4.29 (2H, ABq, $J = 13.4$ Hz), 3.69 (3H, s), 3.46 and 3.37 (2H,	1790, 1724, 1690
4f	ABq, $J = 17.9$ Hz), 1.49 (9H, s) (CDCl ₃); 9.3~9.0 (1H, br s), 7.89 (1H, s), 7.77 (1H, d, $J = 8.2$ Hz), 7.5~7.2 (25H, m), 7.00 (1H, s), 6.89 (1H, s), 5.85 (1H, dd, $J = 4.6$ and 8.2 Hz), 5.05 (1H, d, $J = 4.6$ Hz), 4.25 (2H, s), 3.76 (3H, s), 3.49 and 3.41 (2H, ABq, $J = 17.8$ Hz), 1.50	1790, 1725, 1690
4g	(9H, s) (CDCl ₃); $10.1 \sim 9.7$ (1H, br s), 8.27 (1H, d, $J=8.5$ Hz), 8.05 (1H, s), 7.5 \sim 7.2 (25H, m), 6.95 (1H, s), 6.91 (1H, s), 5.94 (1H, dd, $J=5.0$ and 8.5 Hz), 5.04 (1H, d, $J=5.0$ Hz), 4.67 and 4.34 (2H, ABq, $J=14.2$ Hz), 3.56 and 3.43 (2H, ABq,	1789, 1723, 1689
4h	J=17.2 Hz), 3.31 (3H, s), 1.46 (9H, s) (CDCl ₃ -CD ₃ OD); 7.50~7.15 (25H, m), 7.06 (1H, s), 6.95 (1H, s), 5.99 (1H, d, $J=4.8$ Hz), 5.10 (1H, d, $J=4.8$ Hz), 4.46 (2H, s), 3.71 and 3.58 (2H, ABq, $L=17.6$ Hz), $L=12.6$ (2H, s)	1786, 1717, 1672
4 i	J = 17.6 Hz), 1.52 (9H, 8) (CDCl ₃ - CD ₃ OD); 7.50~7.15 (25H, m), 7.05 (1H, s), 6.94 (1H, s), 6.02 (1H, d, $J = 5.0$ Hz), 5.09 (1H, d, $J = 5.0$ Hz), 4.56 (2H, s), 3.81 (3H, s), 3.73 and 3.57 (2H,	1788, 1717, 1686
4j	ABq, $J=17.6$ Hz), 1.52 (9H, s) (CDCl ₃); 8.9~8.7 (1H, br s), 7.57 (1H, d, $J=8.2$ Hz), 7.5~7.2 (25H, m), 6.98 (1H, s), 6.89 (1H, s), 5.85 (1H, dd, $J=4.8$ and 8.2 Hz), 5.07 (1H, d, $J=4.8$ Hz),	1792, 1725, 1690
4k	4.28 (2H, s), 4.25 (3H, s), 3.40 (2H, br s), 1.30 (9H, s) (CDCl ₃ -CD ₃ OD); 8.51 (1H, s), 7.50~7.15 (25H, m), 7.05 (1H, s), 6.99 (1H, s), 6.02 (1H, d, $J = 5.0$ Hz), 5.13 (1H, d, $J = 5.0$ Hz), 4.16 and 4.00 (2H, ABq, L = 13.7 Hz) 3.67 and 3.51 (2H ABq, $L = 17.6$ Hz) 1.52 (0H s)	1789, 1718, 1686
41	(CDCl_3) ; 8.97 (1H, s), 8.8~8.6 (1H, br s), 7.58 (1H, d, $J=8.8$ Hz), 7.5~7.2 (25H, m), 7.03 (1H, s), 6.97 (1H, s), 5.98 (1H, dd, $J=4.8$ and 8.8 Hz), 7.5~7.2 (25H, $J=4.8$ Hz), 4.58 and 4.53 (2H, ABq, $J=13.7$ Hz), 3.67 and 3.51 (2H, ABq, $J=175$ Hz), 1.50 (0H, c)	1787, 1719, 1690
4m	$(CDCl_3)$; 8.9~8.6 (1H, br s), 7.69 (1H, d, $J=8.8$ Hz), 7.5~7.2 (25H, m), 7.03 (1H, s), 6.97 (1H, s), 6.00 (1H, dd, $J=4.8$ and 8.8 Hz), 5.07 (1H, d, $J=4.8$ Hz), 4.50 (2H, s), 3.68 and 3.53 (2H, ABa, $I=1.75$ Hz), 2.69 (2H, s), 1.50 (0H, s)	1787, 1720, 1690
4n	(CDCl ₃); 8.55 (1H, br s), 8.41 (1H, ddd, $J=1.0$, 1.8 and 4.9 Hz), 7.53 ~7.11 (28H, m), 7.03 (1H, s), 7.01 (1H, ddd, $J=1.0$, 4.9 and 7.4 Hz), 6.90 (1H, s), 5.87 (1H, dd, $J=4.9$ and 8.6 Hz), 5.04 (1H, d, $J=4.9$ Hz), 4.45 (2H, s), 3.56 and 3.42 (2H, ABq, $J=17.2$ Hz), 1.51 (9H, s)	1784, 1717, 1686

Table 3. ¹H NMR and IR spectral data of $4a \sim 4n$.

14, 16, 27 and 44) according to the procedure described for 4a. The spectral data of $4a \sim 4d$, 4h and 4n are listed in Table 3.

Method B: Diphenylmethyl 7β -[(Z)-2-(2-*tert*-Butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-mercapto-3-cephem-4-carboxylate 1-Oxide Silver Salt (5)

A solution of 3 (20.0 g, 20.6 mmol) in methylene chloride (200 ml) was cooled to -30° C. mCPBA

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(80%) (4.89 g, 22.7 mmol) was added to this solution. After being stirred at $-20 \sim -30^{\circ}$ C for 30 minutes, the mixture was diluted with methylene chloride, washed successively with 5% aq NaHCO₃ and brine, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene - EtOAc, $5:1 \rightarrow 3:1$) to yield 18.6 g (91%) of diphenylmethyl 7β -[(Z)-2-(2-tert-butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyminoacetamido]-3-methanesulfonyloxy-3-cephem-4-carboxylate 1-oxide as pale brown froth: ¹H NMR (CDCl₃) δ 8.33 (1H, br s), 7.88 (1H, d, J = 10 Hz), 7.50 \sim 7.15 (25H, m), 7.00 (2H, s), 6.31 (1H, dd, J = 5.0 and 10.1 Hz), 4.55 (1H, d, J = 5.0 Hz), 3.89 and 3.35 (2H, ABq, J = 18.6 Hz), 2.74 (3H, s), 1.50 (9H, s); IR (CHCl₃) cm⁻¹ 3400, 1806, 1725, 1687, 1543, 1510, 1493, 1368, 1154, 1045.

A solution of this sulfoxide (18.6 g, 18.8 mmol) in DMF (150 ml) was cooled to -30° C and treated with sodium hydrosulfide (70%) (3.77 g, 47.1 mmol). After the mixture was stirred at $-20 \sim -30^{\circ}$ C for 1.5 hours, 10% HCl (15 ml) was added to the mixture. This mixture was diluted with water, extracted with EtOAc, washed with brine, dried over anhydrous Na₂SO₄ and evaporated. The residue was dissolved in toluene and concentrated to yield 17.6 g (93%) of diphenylmethyl 7 β -[(Z)-2-(2-tert-butoxycarbonyl-aminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-mercapto-3-cephem-4-carboxylate 1-oxide which contained 8% of toluene: ¹H NMR (CDCl₃) δ 8.41 (1H, br s), 7.84 (1H, d, J=10.0 Hz), 7.61 ~ 7.15 (25H, m), 7.01 (1H, s), 6.90 (1H, s), 6.25 (1H, dd, J=4.8 and 10.0 Hz), 5.12 (1H, br s), 4.50 (1H, d, J=4.8 Hz), 3.67 and 3.27 (2H, ABq, J=18.4 Hz), 1.49 (9H, s); IR (CHCl₃) cm⁻¹ 3396, 1799, 1715, 1686, 1543, 1509, 1493, 1445, 1383, 1369, 1040.

A solution of this mercaptan (92%) (17.5 g, 17.5 mmol) in THF (120 ml) was treated with an aqueous solution of silver nitrate (3.26 g, 19.2 mmol in 16 ml of water) under ice-cooling. After being stirred at ice-bath temperature for 30 minutes, the mixture was diluted with methylene chloride and water. The organic layer was separated, washed with water, dried over anhydrous Na_2SO_4 and evaporated to give 19.7 g of dark yellow froth: Calculated purity of 5 was 92%.

A solution of 23 (1.47 g) in HMPA (3 ml) was added to a solution of 5 (92%) (3.37 g, 3.00 mmol) in HMPA (20 ml). After the mixture was stirred at room temperature for 3 hours, brine and EtOAc were added to the mixture and the resulting precipitate was filtered off. The organic layer was separated, washed with brine six times, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene-EtOAc, 1:1) to yield 1.76 g (56%) of **6e** as pale brown froth.

Compounds 6f, $6i \sim 6m$ and 8 were similarly prepared from 5 with corresponding iodides (7, 22, 32, 33, 36, 41 and 42) according to the procedure described for 6e. The spectral data of 6e, 6f, $6i \sim 6m$ and 8 are listed in Table 4.

Diphenylmethyl 7β -[(Z)-2-(2-tert-Butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-(1H-1,2,4-triazol-3-yl)thiomethylthio-3-cephem-4-carboxylate 1-Oxide (9)

A solution of 8 (11.3 g, 8.83 mmol) in acetone (60 ml) was treated with *p*-toluenesulfonic acid monohydrate (1.68 g, 8.83 mmol) under ice-cooling. After the mixture was stirred at room temperature for 4 hours, 5% aq NaHCO₃ and EtOAc were added to the mixture. The organic layer was separated, washed with brine, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene-EtOAc, $1:1 \rightarrow 1:2$) to yield 2.84g (31%) of 9 as brown froth: ¹H NMR (CDCl₃ - CD₃OD) δ 8.01 (1H, s), 7.4~7.15 (25H, m), 7.03 (1H, s), 6.89 (1H, s), 6.22 (1H, d, J=4.7 Hz), 4.62 (1H, d, J=4.7 Hz), 4.30 (2H, s), 4.06 and 3.58 (2H, ABq, J=17.8 Hz), 1.51 (9H, s); IR (CHCl₃) cm⁻¹ 3380, 3200 (br), 1803, 1720, 1690, 1547, 1510, 1497, 1450, 1372, 1040.

Diphenylmethyl 7β -[(Z)-2-(2-tert-Butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-(4-methyl-4H-1,2,4-triazol-3-yl)thiomethylthio-3-cephem-4-carboxylate 1-Oxide (**6g**)

A solution of 9 (2.71 g, 2.61 mmol) in THF (50 ml) was treated with 1.0 M THF solution of lithium hexamethyldisilazane (2.9 ml) at -78° C. After the mixture was stirred at the same temperature for a few minutes, methyl triflate (0.33 ml, 2.92 mmol) was added to the mixture, which was stirred at -78° C for

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Com- pound No.	¹ H NMR (Solvent, δ)	IR (CHCl ₃) cm ⁻¹ (C=O)
6e	(CDCl ₃); 8.59 (1H, br s), 8.02 (1H, d, $J=10.0$ Hz), 7.79 (1H, s), 7.5~7.2 (25H, m), 6.97 (1H, s), 6.96 (1H, s), 6.25 (1H, dd, $J=4.8$ and 10.0 Hz), 4.48 and 4.22 (2H, ABq, $J=13.7$ Hz), 4.38 (1H, d, $J=4.8$ Hz), 3.97 and 3.47 (2H, ABq, $J=18.8$ Hz) 3.66 (3H s) 1.47 (9H s)	1805, 1722, 1688
6f	(CDCl_3) ; 8.66 (1H, br s), 8.02 (1H, d, $J=9.8\text{Hz}$), 7.91 (1H, s), 7.55~7.20 (25H, m), 6.98 (1H, s), 6.93 (1H, s), 6.24 (1H, dd, $J=4.8\text{and}9.8\text{Hz}$), 4.46 (1H, d, $J=4.8\text{Hz}$), 4.28 (2H, s), 4.12 and 3.49 (2H, ABq, $J=18.4\text{Hz}$), 3.76 (3H, s), 1.46 (9H, s)	1803, 1724, 1689
6g	$(\text{CDCl}_3 - \text{CD}_3\text{OD});$ 8.11 (1H, br s), 7.5~7.2 (25H, m), 7.01 (1H, s), 6.96 (1H, s), 6.26 (1H, d, $J = 4.8 \text{ Hz}),$ 4.74 (1H, d, $J = 4.8 \text{ Hz}),$ 4.70 and 4.12 (2H, ABq, J = 14.6 Hz), 3.85 (2H, s), 3.31 (3H, s), 1.51 (9H, s)	1805, 1722, 1690
6i	(CDCl_3) ; 8.48 (1H, br s), 7.73 (1H, d, $J = 10.2 \text{ Hz}$), 7.50 ~ 7.10 (25H, m), 7.00 (1H, s), 6.97 (1H, s), 6.27 (1H, dd, $J = 4.6$ and 10.2 Hz), 4.50 (1H, d, $J = 4.6 \text{ Hz}$), 4.76 and 4.22 (2H, ABq, $J = 14.2 \text{ Hz}$), 3.89 and 3.73 (2H, ABq, $J = 17.6 \text{ Hz}$), 3.75 (3H, s), 1.48 (0H, s)	1802, 1718, 1686
6j	(CDCl ₃); 8.45 (1H, br s), 7.91 (1H, d, $J=10.0$ Hz), 7.5~7.2 (25H, m), 6.99 (1H, s), 6.95 (1H, s), 6.27 (1H, dd, $J=4.8$ and 10.0 Hz), 4.49 (1H, d, $J=4.8$ Hz), 4.41 and 4.24 (2H, ABq, $J=13.9$ Hz), 4.22 (3H, s), 4.02 and 3.50 (2H, ABq, $J=17.9$ Hz), 4.7 (PH s)	1806, 1725, 1690
6k	(CDCl_3) ; 8.47 (1H, s), 8.45 (1H, br s), 7.96 (1H, d, $J=10.0 \text{ Hz}$), 7.50~7.15 (25H, m), 7.00 (1H, s), 6.98 (1H, s), 6.31 (1H, dd, $J=4.8 \text{ and } 10.0 \text{ Hz}$), 4.49 (1H, d, $J=4.8 \text{ Hz}$), 4.09 and 3.91 (2H, ABq, $J=14.1 \text{ Hz}$), 3.91 and 3.23 (2H, ABq, $J=17.6 \text{ Hz}$), 1.48 (9H, s)	1804, 1718, 1690
61	(CDCI_3) ; 9.00 (1H, s), 8.7~8.45 (1H, br s), 7.86 (1H, d, $J=10.1\text{Hz}$), 7.5~7.2 (25H, m), 7.00 (1H, s), 6.98 (1H, s), 6.27 (1H, dd, $J=4.8$ and 10.1 Hz), 4.84 and 4.29 (2H, ABq, $J=14.2\text{Hz}$), 4.55 (1H, d, $J=4.8\text{Hz}$), 3.98 and 3.74 (2H, ABq, $J=17.9\text{Hz}$), 1.49 (9H, s)	1802, 1718, 1688
бm	(CDCl_3) ; 8.6~8.4 (1H, br s), 7.89 (1H, d, $J=10.2 \text{ Hz}$), 7.5~7.2 (25H, m), 7.00 (1H, s), 6.97 (1H, s), 6.28 (1H, dd, $J=4.6 \text{ and } 10.2 \text{ Hz}$), 4.75 and 4.23 (2H, ABq, $J=14.2 \text{ Hz}$), 4.60 (1H, d, $J=4.6 \text{ Hz}$), 3.97 and 3.76 (2H, ABq, $J=18.2 \text{ Hz}$), 2.68 (3H, s), 1.49 (9H, s)	1802, 1718, 1686
8	$(CDCl_3)$; 8.30 (1H, br s), 8.04 (1H, d, $J=10.1$ Hz), 7.90 (1H, s), 7.5~7.05 (41H, m), 6.97 (1H, s), 6.19 (1H, dd, $J=4.6$ and 10.1 Hz), 4.22 and 4.16 (2H, ABq, $J=13.7$ Hz), 4.19 (1H, d, $J=4.6$ Hz), 3.82 and 3.19 (2H, ABq, $J=18.5$ Hz), 1.49 (9H, s)	1804, 1725, 1689

Table 4. ¹H NMR and IR spectral data of $6e \sim 6g$, $6i \sim 6m$ and 8.

30 minutes. The reaction was quenched by 10% HCl (2.1 ml). The mixture was diluted with water and extracted with EtOAc. The extract was washed with brine, dried over anhydrous Na_2SO_4 and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; EtOAc) to yield 730 mg (27%) of **6g** as pale brown froth. The spectral data of **6g** are listed in Table 4.

Diphenylmethyl 7β -[(Z)-2-(2-tert-Butoxycarbonylaminothiazol-4-yl)-2-triphenylmethoxyiminoacetamido]-3-(1-methyl-1H-1,2,4-triazol-5-yl)thiomethylthio-3-cephem-4-carboxylate (4e)

A solution of **6e** (1.73 g, 1.64 mmol) in DMF (15 ml) was treated with phosphorus trichloride (0.41 ml, 4.08 mmol) at -20° C. After being stirred at the same temperature for 20 minutes, the mixture was poured into a cold mixture of 5% aq NaHCO₃ (*ca.* 42 ml) and EtOAc. The organic layer was separated, washed with brine four times, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent: toluene - EtOAc, 2:1) to yield 1.60 g (94%) of **4e** as brown froth.

Compounds 4f, 4g and 4i \sim 4m were similarly prepared according to the procedure described for 4e. The spectral data of 4e \sim 4g and 4i \sim 4m are listed in Table 3.

Com- pound No.	¹ H NMR (D ₂ O + NaHCO ₃ , δ)	IR (KBr) cm ⁻¹ (C=O)
2a	8.09 (1H, s), 6.99 (1H, s), 5.83 (1H, d, $J=4.6$ Hz), 5.24 (1H, d, $J=4.6$ Hz), 4.22 and 4.12 (2H ABa $I=14.0$ Hz) 4.11 (3H s) 3.81 and 3.50 (2H ABa $I=17.4$ Hz)	1770, 1650
2b	7.84 (1H, s), 6.99 (1H, s), 5.82 (1H, d, $J=4.8$ Hz), 5.23 (1H, d, $J=4.8$ Hz), 4.25 and 4.18 (2H, ABg , $L=13.7$ Hz), 4.17 (2H, c), 2.80 and 2.52 (2H, ABg, $L=17.2$ Hz)	1765, 1660
2c	7.95 (1H, s), 6.98 (1H, s), 5.84 (1H, d, $J=4.5$ Hz), 5.23 (1H, d, $J=4.5$ Hz), 4.27 and 4.14 (2H, s), 6.98 (1H, s), 5.84 (1H, d, $J=4.5$ Hz), 5.23 (1H, d, $J=4.5$ Hz), 4.27 and	1765, 1660
2đ	4.14 (2ft, ABd, $J = 17.4$ Hz), 4.08 (3ft, s), 5.78 and 5.55 (2ft, ABd, $J = 17.4$ Hz) 8.40 (1H, s), 6.98 (1H, s), 5.83 (1H, d, $J = 4.3$ Hz), 5.21 (1H, d, $J = 4.3$ Hz), 4.42 (2H, s), 8.70 ± 1.2 54 (2H, ABd, $J = 17.4$ Hz)	1765, 1655
2e	5.79 and 5.54 (2H, ABq, $J = 17.4$ Hz) 8.03 (1H, s), 6.98 (1H, s), 5.83 (1H, d, $J = 4.8$ Hz), 5.21 (1H, d, $J = 4.8$ Hz), 4.50 and	1765, 1655
2f	4.44 (2H, ABq, $J = 14.1$ Hz), 3.85 (3H, s), 3.81 and 3.54 (2H, ABq, $J = 17.4$ Hz) 8.36 (1H, s), 6.99 (1H, s), 5.82 (1H, d, $J = 4.7$ Hz), 5.24 (1H, d, $J = 4.7$ Hz), 4.40 (2H, s),	1765, 1660
2g	5.89 (3H, s), 3.83 and 3.55 (2H, ABq, $J = 17.2$ Hz) 8.50 (1H, s), 7.00 (1H, s), 5.83 (1H, d, $J = 4.9$ Hz), 5.19 (1H, d, $J = 4.9$ Hz), 4.50 and	1767, 1655
2h	4.35 (2H, ABq, $J = 13.4$ Hz), 3.80 and 3.53 (2H, ABq, $J = 17.5$ Hz), 3.69 (3H, s) 6.99 (1H, s), 5.83 (1H, d, $J = 4.6$ Hz), 5.18 (1H, d, $J = 4.6$ Hz), 4.43 and 4.38 (2H, ABq,	1765, 1650
2i	J=13.7 Hz), 3.65 and 3.47 (2H, ABq, $J=17.4$ Hz) 6.98 (1H, s), 5.83 (1H, d, $J=4.9$ Hz), 5.24 (1H, d, $J=4.9$ Hz), 4.63 and 4.58 (2H, ABq,	1765, 1660
2j	J=13.8 Hz), 4.00 (3H, s), 3.90 and 3.59 (2H, ABq, $J=17.4$ Hz) 6.98 (1H, s), 5.82 (1H, d, $J=4.7$ Hz), 5.26 (1H, d, $J=4.7$ Hz), 4.50 (2H, s), 4.36 (3H, s),	1767, 1660
2k	3.88 and 3.58 (2H, ABq, $J = 17.3$ Hz) 8.76 (1H, s), 6.97 (1H, s), 5.83 (1H, d, $J = 4.7$ Hz), 5.25 (1H, d, $J = 4.7$ Hz), 4.48 and	1760, 1655
21	4.37 (2H, ABq, $J = 14.1$ Hz), 3.88 and 3.58 (2H, ABq, $J = 17.4$ Hz) 9.41 (1H, s), 6.98 (1H, s), 5.83 (1H, d, $J = 4.7$ Hz), 5.24 (1H, d, $J = 4.7$ Hz), 4.64 and	1765, 1665
2m	4.57 (2H, ABq, $J = 14.1$ Hz), 3.89 and 3.60 (2H, ABq, $J = 17.4$ Hz) 6.97 (1H, s), 5.83 (1H, d, $J = 4.8$ Hz), 5.23 (1H, d, $J = 4.8$ Hz), 4.57 and 4.51 (2H, ABq,	1772, 1668
2n	J=14.0 Hz), 3.87 and 3.58 (2H, ABq, $J=17.4$ Hz), 2.72 (3H, s) 8.39 (1H, br d, $J=4.9$ Hz), 7.74 (1H, ddd, $J=1.6$, 7.5 and 8.0 Hz), 7.46 (1H, br d,	1760, 1665
	J=8.0 Hz), 7.23 (1H, ddd, $J=0.8$, 4.9 and 7.5 Hz), 6.96 (1H, s), 5.80 (1H, d, $J=4.6$ Hz), 5.16 (1H, d, $J=4.6$ Hz), 4.49 and 4.43 (2H, ABq, $J=14.0$ Hz), 3.78 and 3.55 (2H, ABq, $J=17.2$ Hz)	

Table 5. ¹H NMR and IR spectral data of $2a \sim 2n$.

Deprotection of 4: General Procedure Illustrated with the Preparation of 7β -[(Z)-2-(2-Aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-(2-methyl-2H-1,2,3-triazol-4-yl)thiomethylthio-3-cephem-4-carboxylic Acid (2b)

A solution of AlCl₃ (1.43 g, 10.8 mmol) in anisole (5 ml) was added dropwise to a solution of **4b** (1.39 g, 1.34 mmol) in anisole (5 ml) and nitromethane (20 ml) at $-30 \sim -40^{\circ}$ C. After the mixture was stirred at the same temperature for an hour, 11 ml of 1 N HCl, water and EtOAc were added to the mixture. The aqueous layer was separated and the organic layer was re-extracted with water. The combined aqueous layer was chromatographed on an Diaion HP-20 column (eluent; methanol-water, 4:1). After the concentration, the resulting precipitate was collected by filtration, washed with EtOAc and dried *in vacuo* to give 534 mg (75%) of **2b** as pale yellow powder.

The spectral data of various derivatives $2a \sim 2n$ are listed in Table 5.

1*H*-1,2,3-Triazol-4-ylthiomethyl Thioacetate (11)

A suspension of **10** (purchased from Dynamit Nobel Aktiengesellschaft) (38.0 g, 0.31 mol) in DMF (150 ml) was treated with chloromethyl thioacetate⁷⁾ (37.4 g, 0.30 mol) at -20° C. After being stirred at room temperature for 2 hours, the mixture was poured into water, extracted with EtOAc. The organic layer was washed with brine four times, dried over anhydrous Na₂SO₄ and evaporated. The resulting crystalline residue was washed with hexane and dried *in vacuo*. Recrystallization from ether gave 38.9 g (69%) of **11** as white crystals: MP 88~89°C; ¹H NMR (CDCl₃) δ 7.73 (1H, s), 6.3 (1H, br s), 4.37 (2H, s), 2.36 (3H, s); IR (CHCl₃) cm⁻¹ 3430, 3152, 1693; MS *m*/*z* 189 (M⁺).

Anal Calcd for C₅H₇N₃OS₂: C 31.73, H 3.73, N 22.20, S 33.88.

Found:

C 31.65, H 3.69, N 22.44, S 33.62.

1-Methyl-1*H*-1,2,3-triazol-4-ylthiomethyl Thioacetate (12), 2-Methyl-2*H*-1,2,3-triazol-4-ylthiomethyl Thioacetate (13) and 1-Methyl-1*H*-1,2,3-triazol-5-ylthiomethyl Thioacetate (14)

Methylation of 11 with Diazomethane

A solution of 11 (5.80 g, 30.7 mmol) in THF (50 ml) was treated with an ether solution of diazomethane under ice-cooling. After being stirred at the same temperature for an hour, the mixture was concentrated. The resulting residue was purified by Lobar column chromatography (eluent; toluene-EtOAc, $10:1\rightarrow1:1\rightarrow1:2$) to afford 306 mg (5%) of 12 as white crystals, 3.77 g (61%) of 13 as a colorless oil and 429 mg (7%) of 14 as white crystals.

Methylation of 11 with Methyl Triflate

A solution of 11 (6.00 g, 31.8 mmol) in THF (30 ml) was treated with 1.0 M THF solution of lithium hexamethyldisilazane (35 ml) at -78° C. A few minutes later, methyl triflate (4.0 ml, 35.3 mmol) was added to the mixture. After the mixture was stirred at -78° C for 2 hours, 26 ml of 10% HCl was added to the mixture, which was diluted with water, extracted with EtOAc, washed with brine, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by Lobar column chromatography (eluent; toluene-EtOAc, 1:2) to afford 884 mg (14%) of 12, 175 mg (3%) of 13 and 2.22 g (34%) of 14.

12: MP 71 ~ 72°C; ¹H NMR (CDCl₃) δ 7.59 (1H, s), 4.33 (2H, s), 4.11 (3H, s), 2.35 (3H, s); IR (CHCl₃) cm⁻¹ 1691, 1434, 1354; MS *m*/*z* 203 (M⁺).

13: ¹H NMR (CDCl₃) δ 7.56 (1H, s), 4.30 (2H, s), 4.20 (3H, s), 2.35 (3H, s); IR (CHCl₃) cm⁻¹ 1691, 1446, 1369; MS *m*/*z* 203 (M⁺).

Anal Calcd for C₆H₉N₃OS₂: C 35.45, H 4.46, N 20.67, S 31.55. Found: C 35.29, H 4.55, N 20.66, S 31.25.

14: MP 37~38°C; ¹H NMR (CDCl₃) δ 7.77 (1H, s), 4.13 (5H, s), 2.32 (3H, s); IR (CHCl₃) cm⁻¹ 1698, 1429, 1355; MS m/z 203 (M⁺).

1-Triphenylmethyl-1H-1,2,4-triazol-3-ylthiomethyl Thioacetate (16)

A solution of 15 (2.23 g, 22.1 mmol) in DMF (30 ml) was treated with sodium hydride (60% in oil) (840 mg, 21.0 mmol) and stirred at room temperature for 10 minutes. After the mixture was cooled to -60° C, a solution of chloromethyl thioacetate⁷¹ (2.50 g, 20.1 mmol) in DMF (5 ml) was added to the reaction mixture. After the mixture was stirred at $-50 \sim -60^{\circ}$ C for 20 minutes, trityl chloride (6.70 g, 24.0 mmol) and pyridine (1.94 ml, 24.0 mmol) were added to the mixture, which was stirred at ice-bath temperature for 28 hours. The reaction mixture was diluted with water and extracted with EtOAc. The extract was washed with brine four times, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene - EtOAc, 20:1) and crystallized from ether to yield 3.37 g (39%) of **16** as white crystals: MP 124~125°C; ¹H NMR (CDCl₃) δ 7.90 (1H, s), 7.4~7.3 (9H, m), 7.2~7.1 (6H, m), 4.50 (2H, s), 2.32 (3H, s); IR (CHCl₃) cm⁻¹ 1690, 1490, 1472, 1444, 1383, 1352.

HRMS Calcd for $C_{24}H_{21}N_3OS_2$ (M⁺): 431.1126. Found: m/z 431.1108 (M⁺).

3-p-Methoxybenzylthio-1H-1,2,4-triazole (17)

To a solution of 15 (10.1 g 0.10 mol) and *p*-methoxybenzyl chloride (17.2 g, 0.11 mol) in methylene chloride (50 ml), $1 \times aq$ NaOH (105 ml) and tetra-*n*-butylammonium bromide (750 mg, 2.33 mmol) were added. After the mixture was stirred at room temperature for 16 hours, the organic layer was separated, washed with brine, dried over anhydrous Na₂SO₄ and evaporated. The residue was crystallized from

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toluene to afford 16.7 g (76%) of **17** as white crystals: MP 100~101°C; ¹H NMR (CDCl₃) δ 8.13 (1H, s), 7.8~7.0 (1H, br s), 7.27~7.23 and 6.83~6.79 (4H, AA'BB'), 4.32 (2H, s), 3.77 (3H, s); IR (CHCl₃) cm⁻¹ 3440, 3120 (br), 1611, 1512, 1485, 1465, 1441, 1302; MS *m/z* 221 (M⁺). *Anal* Calcd for C₁₀H₁₁N₃OS: C 54.28, H 5.01, N 18.99, S 14.49. Found: C 54.26, H 5.09, N 18.86, S 14.49. <u>3-p-Methoxybenzylthio-1-methyl-1*H*-1,2,4-triazole(**18**) and 5-p-Methoxybenzylthio-1-methyl-1*H*-1,2,4-striazole (**19**) Compound **17** was treated with diazomethane as described for the preparation of **12~14** to yield **18** and **19** as colorless oils.</u>

18: ¹H NMR (CDCl₃) δ 7.99 (1H, s), 7.34~7.30 and 6.85~6.80 (4H, AA'BB'), 4.31 (2H, s), 3.87 (3H, s), 3.78 (3H, s); IR (CHCl₃) cm⁻¹ 1612, 1512, 1465, 1440, 1421, 1356, 1302.

HRMS Calcd for C₁₁H₁₃N₃OS (M⁺): 235.0779.

Found: *m*/*z* 235.0791 (M⁺).

19: ¹H NMR (CDCl₃) δ 7.88 (1H, s), 7.24~7.19 and 6.84~6.80 (4H, AA'BB'), 4.34 (2H, s), 3.79 (3H, s), 3.63 (3H, s); IR (CHCl₃) cm⁻¹ 1611, 1511, 1476, 1464, 1440, 1360, 1302.

HRMS Calcd for $C_{11}H_{13}N_3OS$ (M⁺): 235.0779. Found: m/z 235.0782 (M⁺).

3-Chloromethylthio-1-methyl-1H-1,2,4-triazole (20)

A solution of 18 (3.55 g, 15.1 mmol) in methylene chloride (30 ml) and methanol (30 ml) was treated with silver perchlorate (90%) (4.17 g, 18.1 mmol). After being stirred at room temperature for 1.5 hours, the mixture was diluted with methanol. The precipitate was collected by filtration and dried *in vacuo*. This white solid was suspended in DMF (30 ml), and bromochloromethane (30 ml) and lithium chloride (98%) (1.96 g, 45.3 mmol) were added to the suspension, which was stirred at room temperature for 20 hours. Brine and EtOAc were added to the reaction mixture and filtered. The organic layer was separated, washed with brine three times, dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was purified by column chromatography on silica gel (eluent; toluene - EtOAc, 2:1) to yield 1.05 g (43%) of **20** as a colorless oil: ¹H NMR (CDCl₃) δ 8.06 (1H, s), 5.21 (2H, s), 3.93 (3H, s); IR (CHCl₃) cm⁻¹ 1509, 1471, 1424, 1392, 1359.

HRMS Calcd for $C_4H_6ClN_3S$ (M⁺): 162.9971. Found: m/z 162.9984 (M⁺).

5-Chloromethylthio-1-methyl-1*H*-1,2,4-triazole (21)

Compound 21 was prepared from 19 as described for the preparation of 20: ¹H NMR (CDCl₃) δ 7.96 (1H, s), 5.19 (2H, s), 3.87 (3H, s); IR (CHCl₃) cm⁻¹ 1479, 1395, 1360.

HRMS Calcd for $C_4H_6ClN_3S$ (M⁺): 162.9971.

Found: *m/z* 162.9972 (M⁺).

3-Iodomethylthio-1-methyl-1*H*-1,2,4-triazole (22)

A solution of **20** (981 mg, 6.00 mmol) in acetone (10 ml) was treated with sodium iodide (1.78 g, 12.0 mmol) and stirred at 50°C for 3 hours. The reaction mixture was diluted with water and extracted with EtOAc. The extract was washed with water, dried over anhydrous Na₂SO₄ and evaporated to afford 1.50 g of crude **22** as a yellow oil. This oil was employed for the reaction with silver salt **5** without purification: ¹H NMR (CDCl₃) δ 8.07 (1H, s), 4.75 (2H, s), 3.94 (3H, s).

5-Iodomethylthio-1-methyl-1*H*-1,2,4-triazole (23)

Compound 23 was prepared from 21 as described for the preparation of 22: ¹H NMR (CDCl₃) δ 7.98 (1H, s), 4.71 (2H, s), 3.84 (3H, s).

3-Chloromethythio-1-triphenylmethyl-1H-1,2,4-triazole (24)

A solution of 15 (10.0 g, 99.0 mmol) in DMF (100 ml) was treated with sodium hydride (60% in oil) (3.96 g,99.0 mmol) under ice-cooling. After the mixture was stirred at the same temperature for 10 minutes, bromochloromethane (100 ml) was added to the mixture, which was stirred at room temperature for 15

hours. The reaction mixture was diluted with water and extracted with EtOAc. The extract was washed with brine four times, dried over anhydrous Na_2SO_4 and evaporated. The resulting residue was dissolved in DMF (100 ml) and cooled to ice-bath temperature. Trityl chloride (27.6 g, 99.0 mmol) and triethylamine (13.8 ml, 99.0 mmol) were added to the solution, which was stirred at the same temperature for 30 minutes. The reaction mixture was diluted with water and extracted with EtOAc. The extract was washed with brine four times, dried over anhydrous Na_2SO_4 and evaporated. The resulting residue was crystallized from ether to give 20.2 g (52%) of **24** as white crystals: MP 121 ~ 122°C; ¹H NMR (CDCl₃) δ 7.95 (1H, s), 7.4~7.3 (9H, m), 7.2~7.1 (6H, m), 5.18 (2H, s); IR (CHCl₃) cm⁻¹ 1599, 1492, 1472, 1445, 1389, 1365, 1353, 1325.

HRMS Calcd for $C_{22}H_{18}ClN_3S$ (M⁺): 391.0910. Found: m/z 391.0885 (M⁺).

3-Iodomethylthio-1-triphenylmethyl-1H-1,2,4-triazole (7)

Compound 7 was prepared from 24 as described for the preparation of 22: ¹H NMR (CDCl₃) δ 7.96 (1H, s), 7.4~7.3 (9H, m), 7.2~7.1 (6H, m), 4.70 (2H, s).

5-Mercapto-1*H*-tetrazole (26)

A solution of 25 (10.0 g, 45.0 mmol), which was similarly prepared as described in the literature,⁸⁾ in trifluoroacetic acid (100 ml) and anisole (20 ml) was heated at 80°C for 2 hours. The reaction mixture was concentrated and the crystalline residue was washed with toluene to obtain 4.33 g (94%) of 26 as white crystals: MP 198~200°C (dec); ¹³C NMR (CD₃OD) δ 167.1; IR (KBr) cm⁻¹ 3430 (br), 3040 (br), 2840, 1513, 1346; MS m/z 102 (M⁺).

Anal Caled for CH₂N₄S: C 11.76, H 1.97, N 54.86, S 31.40. Found: C 11.84, H 2.05, N 54.60, S 31.33.

1H-Tetrazol-5-ylthiomethyl Thioacetate (27)

Compound 27 was similarly prepared from 26 as described for the preparation of 16: MP 90°C; ¹H NMR (CDCl₃) δ 9.0 ~ 8.0 (1H, br s), 4.68 (2H, s), 2.42 (3H, s); IR (CHCl₃) cm⁻¹ 3072 (br), 1692, 1500, 1356.

HRMS Calcd for $C_4H_6N_4OS_2$ (M⁺): 189.9983. Found: m/z 189.9971 (M⁺).

 $(29) \frac{5-p-Methoxybenzylthio-1-methyl-1H-tetrazole(28), 5-p-Methoxybenzylthio-2-methyl-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazole(28), 5-p-Methoxybenzylthio-2H-tetrazol$

Compound 25 was treated with diazomethane as described for the preparation of $12 \sim 14$ to yield 28 as white crystals and 29 as a colorless oil.

28: MP 68~69°C; ¹H NMR (CDCl₃) δ 7.31~7.27 and 6.86~6.82 (4H, AA'BB'), 4.49 (2H, s), 3.80 (3H, s), 3.79 (3H, s); IR (CHCl₃) cm⁻¹ 1613, 1513, 1465, 1305; MS *m/z* 236 (M⁺).

 Anal Calcd for C₁₀H₁₂N₄OS:
 C 50.83, H 5.12, N 23.71, S 13.57.

 Found:
 C 50.84, H 5.16, N 23.55, S 13.36.

29: ¹H NMR (CDCl₃) δ 7.34~7.30 and 6.85~6.81 (4H, AA'BB'), 4.38 (2H, s), 4.29 (3H, s), 3.78 (3H, s); IR (CHCl₃) cm⁻¹ 1611, 1512, 1390, 1324, 1303; MS *m*/*z* 236 (M⁺).

Anal Calcd for $C_{10}H_{12}N_4OS$:C 50.83, H 5.12, N 23.71, S 13.57.Found:C 51.01, H 5.24, N 23.69, S 13.31.

 $\frac{5 - \text{Chloromethylthio-1-methyl-1}H - \text{tetrazole (30) and 5-Chloromethylthio-2-methyl-2}H - \text{tetrazole (31)}}{\text{Compound 30 and 31 were prepared from 28 and 29 as described for the preparation of 20.} } \\ 30: MP 55 ~ 56^{\circ}\text{C}; {}^{1}\text{H NMR (CDCl}_{3}) \delta 5.29 (2\text{H, s}), 4.03 (3\text{H, s}); IR (CHCl_{3}) \text{ cm}^{-1} 1467, 1408, 1384. \\ \text{HRMS Calcd for C}_{3}\text{H}_{5}\text{ClN}_{4}\text{S} \quad (M^{+}): 163.9923. \\ \text{Found: } m/z & 163.9926 (M^{+}). \\ 31: {}^{1}\text{H NMR (CDCl}_{3}) \delta 5.23 (2\text{H, s}), 4.37 (3\text{H, s}); IR (CHCl_{3}) \text{ cm}^{-1} 1440, 1422, 1410, 1395, 1325. \\ \text{HRMS Calcd for C}_{3}\text{H}_{5}\text{ClN}_{4}\text{S} \quad (M^{+}): 163.9923. \\ \end{array}$

Found: *m/z* 163.9939 (M⁺).

5-Chloromethylthio-1,2,3-thiadiazole (35), 2-Chloromethylthio-1,3,4-thiadiazole (39) and 2-Chloromethylthio-5-methyl-1,3,4-thiadiazole (40)

Compound 35, 39 and 40 were similarly prepared from 34,⁹⁾ 37 and 38 (37, 38; purchased from Toyo Kasei Kogyo Co., Ltd.) as described for the preparation of 24.

35: ¹H NMR (CDCl₃) δ 8.69 (1H, s), 4.93 (2H, s); IR (CHCl₃) cm⁻¹ 1419, 1395; MS m/z 166 (M⁺).
 Anal Calcd for C₃H₃ClN₂S₂: C 21.62, H 1.81, Cl 21.27, N 16.81, S 38.48.
 Found: C 21.49, H 2.02, Cl 21.54, N 16.98, S 38.25.

39: ¹H NMR (CDCl₃) δ 9.16 (1H, s), 5.32 (2H, s); IR (CHCl₃) cm⁻¹ 1389, 1373; MS m/z 166 (M⁺).
 Anal Calcd for C₃H₃ClN₂S₂: C 21.62, H 1.81, Cl 21.27, N 16.81, S 38.48.
 Found: C 21.58, H 2.01, Cl 21.31, N 16.84, S 38.59.

40: ¹H NMR (CDCl₃) δ 5.24 (2H, s), 2.79 (3H, s); IR (CHCl₃) cm⁻¹ 1430, 1392, 1380. HRMS Calcd for C₄H₅ClN₂S₂ (M⁺): 179.9583. Found: *m*/*z* 179.9591 (M⁺).

5-Iodomethylthio-1,2,3-thiadiazole (36), 2-Iodomethylthio-1,3,4-thiadiazole (41) and 2-Iodomethylthio-5-methyl-1,3,4-thiadiazole (42)

Compound 36, 41 and 42 were prepared from 35, 39 and 40 as described for the preparation of 22. 36: ¹H NMR (CDCl₃) δ 8.62 (1H, s), 4.53 (2H, s). 41: ¹H NMR (CDCl₃) δ 9.14 (1H, s), 4.85 (2H, s).

42: ¹H NMR (CDCl₃) δ 4.78 (1H, s), 2.79 (3H, s).

Pyridin-2-ylthiomethyl Thioacetate (44)

Compound 44 was similarly prepared from 43 as described for the preparation of 16: ¹H NMR (CDCl₃) δ 8.47 (1H, ddd, $J_{5,6}$ =4.9 Hz, $J_{4,6}$ =1.8 Hz, $J_{3,6}$ =1.0 Hz), 7.51 (1H, ddd, $J_{4,5}$ =7.4 Hz, $J_{3,4}$ =8.0 Hz, $J_{4,6}$ =1.8 Hz), 7.18 (1H, ddd, $J_{3,4}$ =8.0 Hz, $J_{3,5}$ =J_{3,6}=1.0 Hz), 7.03 (1H, ddd, $J_{4,5}$ =7.4 Hz, $J_{5,6}$ =4.9 Hz, $J_{3,5}$ =1.0 Hz), 4.65 (2H, s), 2.35 (3H, s); IR (CHCl₃) cm⁻¹ 1686, 1576, 1556, 1452, 1414, 1353; MS m/z 199 (M⁺).

Anal Calcd for $C_8H_9NOS_2$:C 48.21, H 4.55, N 7.03, S 32.18.Found:C 48.25, H 4.61, N 7.03, S 32.04.

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